

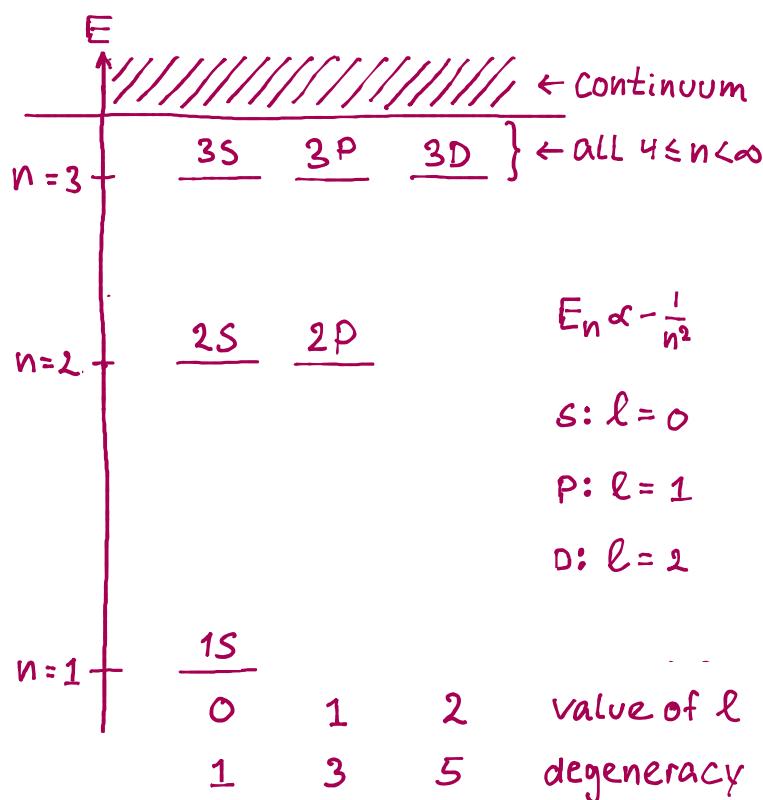
Atom-Light Interaction: Multi-Level Atoms

Starting point – the Hydrogen atom

$$H_A = \frac{p^2}{2m} - \frac{1}{4\pi\epsilon_0} \frac{e^2}{|\vec{r}|}$$

$$V_{ext}(\vec{r}, \vec{R}, t) = -e\vec{r} \cdot \vec{E}(\vec{R}, t)$$

\vec{r} : relative \vec{R} : center-of-mass



Note: Frequencies for transitions $n \rightarrow n'$, $n'' \rightarrow n'''$ are very different \rightarrow near-resonant approx. with a single transition frequency $\omega \sim \omega_0$

Levels $|nl\rangle$ are generally degenerate with respect to the quantum number m , so we cannot isolate a 2-level system only through its transition frequency.

We must therefore consider **Selection Rules**

Interaction matrix element

$$\langle n'l'm' | V_{ext} | nl'm \rangle \propto \int_{-\infty}^{\infty} dr^3 \phi_{n'l'm'}^*(\vec{r}) \vec{r} \cdot \phi_{nl'm}(\vec{r})$$

Wavefunction parity is even/odd depending on l

$$\phi_{nl'm}(\vec{r}) = (-1)^l \phi_{nl'm}(-\vec{r})$$

$\rightarrow \langle |V| \rangle$ can be non-zero only if $(l-l')$ is odd.

This is the **Parity Selection Rule !**

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And thus in the RWA we get (use $(Y_e^m)^* = (-1)^{lm} Y_e^{-m}$)

$$V_{21} \propto \langle l'm' | Y_1^q e^{-i\omega t} | lm \rangle$$

$$V_{12} \propto \langle lm | (-1)^q Y_1^{-q} e^{i\omega t} | l'm' \rangle$$



Contracting the factor $(-1)^q$

$$V_{21} \propto \int d\Omega (Y_e^{m'})^* Y_1^q Y_e^m \propto \langle 1, q; lm | l'm' \rangle$$

$$V_{12} \propto \int d\Omega (Y_e^m)^* Y_1^{-q} Y_e^{m'} \propto \langle 1, -q; l'm' | lm \rangle$$



Clebsch-Gordan coefficients

Next: We can understand this as conservation of angular momentum when a photon is absorbed or emitted



Selection Rules for Electric Dipole Transitions

Reminder: Addition of Angular Momenta

Let $\vec{J} = \vec{j}_1 + \vec{j}_2 \rightarrow$ eigenstates $\left\{ \begin{array}{l} |j_1 m_1\rangle \\ |j_2 m_2\rangle \\ |jm\rangle \end{array} \right.$

We can write $|jm\rangle$ in the basis $|j_1 m_1\rangle |j_2 m_2\rangle$

$$\begin{aligned} |jm\rangle &= \underbrace{\sum_{m_1, m_2} |j_1 m_1; j_2 m_2\rangle}_{\text{identity}} \langle j_1 m_1; j_2 m_2 | jm \rangle \\ &= \sum_{m_1, m_2} \langle j_1 m_1; j_2 m_2 | jm \rangle |j_1 m_1; j_2 m_2\rangle \end{aligned}$$

↑
Clebsch-Gordan coefficients

CG's are non-zero when

Conservation of Angular Momentum

$$|j_1 - j_2| \leq j \leq j_1 + j_2$$

$$m_1 + m_2 = m$$

Going back to the matrix element, $V_{21} \neq 0$

when $|1q\rangle$ combined w/ $|lm\rangle$ is consistent w/ $|l'm'\rangle$

↑
"photon" AM

↑
ground state AM

↑
excited state AM

Atom-Light Interaction: Multi-Level Atoms

The corresponding Selection Rules are

$$\ell' - \ell = 0, \pm 1, \quad m' - m = q, \quad q = 0, \pm 1$$

Combining with the Parity Rule, this gives us the

Electric Dipole Selection Rules

$$\ell' - \ell = \pm 1, \quad m' - m = q, \quad q = 0, \pm 1$$

End 02-08-2021 / Begin 02-10-2021

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Remarkably

- * These selection rules generalize to complex many-electron atoms, and after we include both electron and nuclear spins in the theory.
- * From a physics perspective, this reflects the conservation of angular momentum in rotationally invariant systems, and therefore transitions that do not conserve angular momentum are forbidden
- * To find the Clebsch-Gordan coefficients for different transitions we would need to use the Wigner-Eckart theorem, the proof of which is beyond this course.

General ED Selection Rules

$$\Delta L = \pm 1 \quad \vec{L} : \text{total e orbital A. M.}$$

$$\Delta F = 0, \pm 1 \quad \vec{F} : \text{total orbital + spin A. M.}$$

$$\Delta m_F = q = 0, \pm 1 \quad q : \text{polarization of EM field}$$

Clebsch-Gordan coefficients ($E_{F'm'_F} > E_{F,m_F}$)

$$\langle F'm'_F | V | F, m_F \rangle \propto \langle 1, q; F, m_F | F'm'_F \rangle$$

$$\langle F, m_F | V | F'm'_F \rangle \propto \langle 1, -q; F'm'_F | F, m_F \rangle$$

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Hydrogen atom

$1S - 2S$: forbidden

$1S - 2P$: allowed

Total spin: $\vec{F} = \vec{j} + \vec{l}$, $\vec{j} = \vec{L} + \vec{s}$

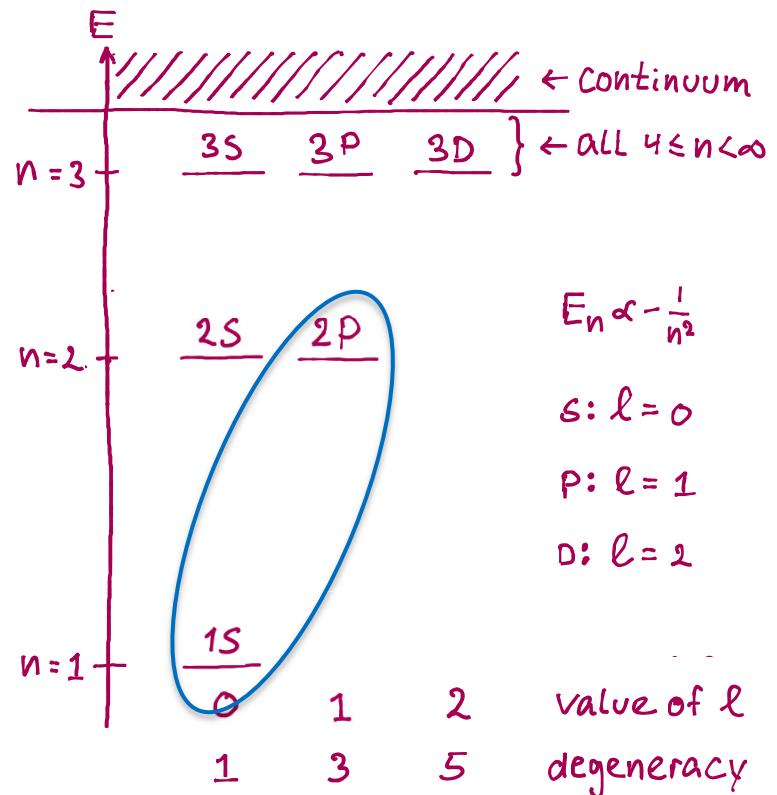
nuclear orbital electron spin

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↑
nuclear ↑
orbital ↑
electron spin

Atom-Light Interaction: Multi-Level Atoms

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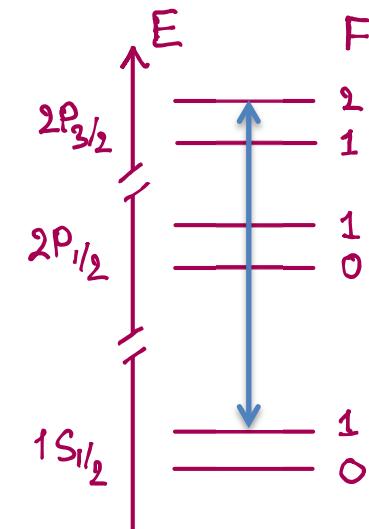
$1S$ State:

$$J = 1/2, F = 0, 1$$

$2P$ State:

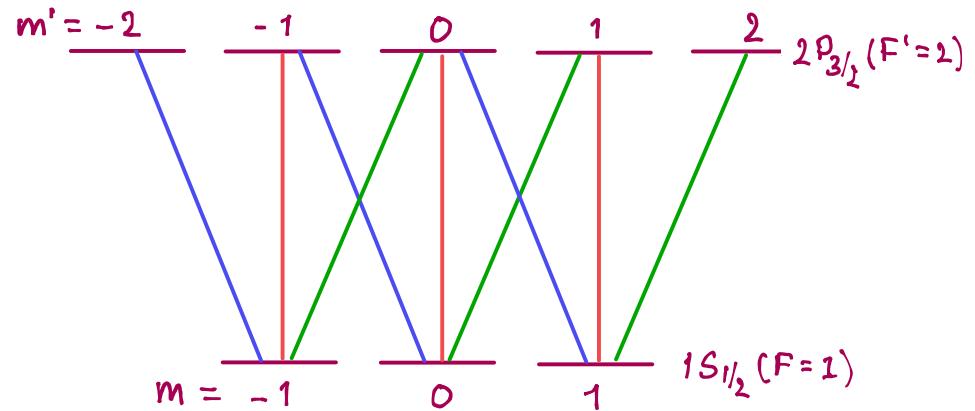
$$J = 1/2, F = 0, 1$$

$$J = 3/2, F = 1, 2$$



Level diagram for transitions

$$1S_{1/2}(F=1) \rightarrow 2P_{3/2}(F'=2)$$



Polarization: $| q=0 / q=1 \setminus q=-1$

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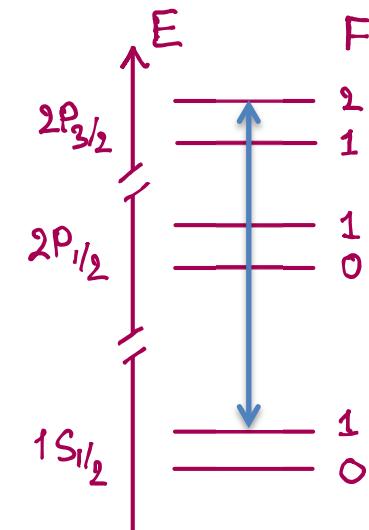
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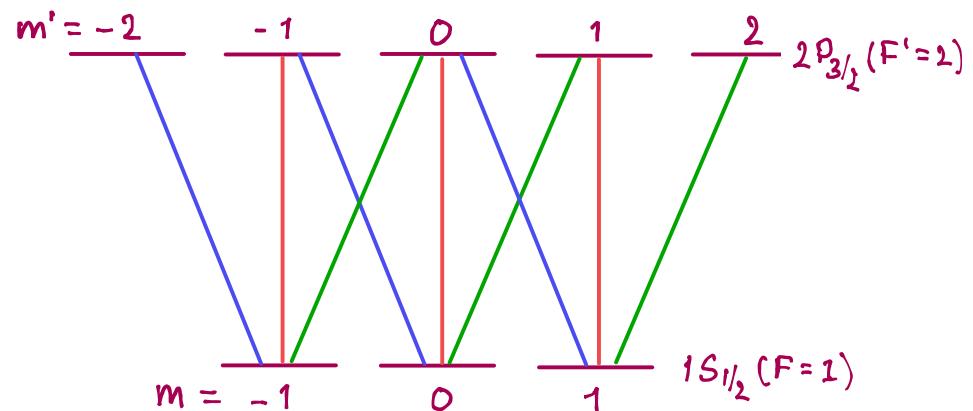
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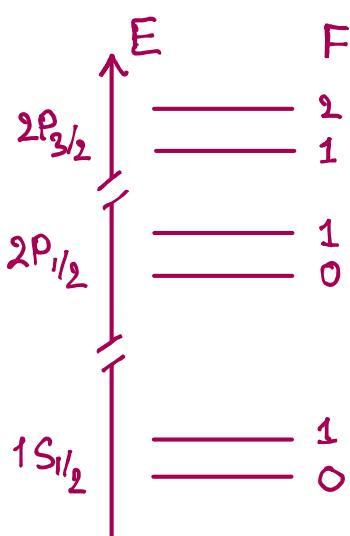
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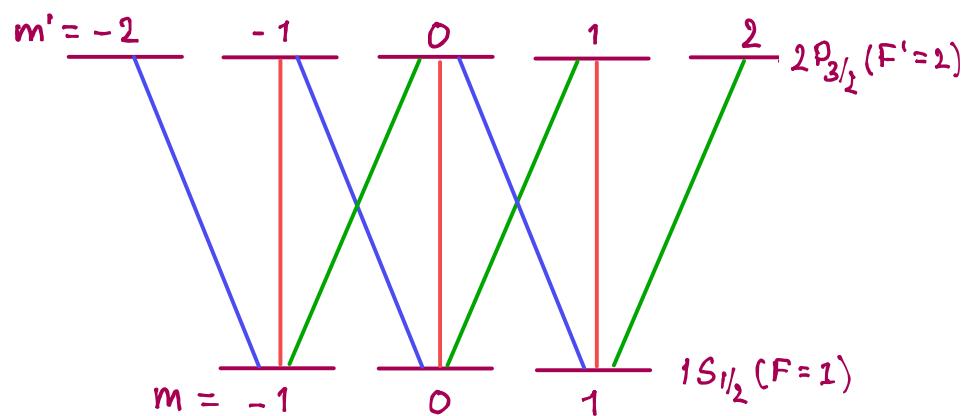
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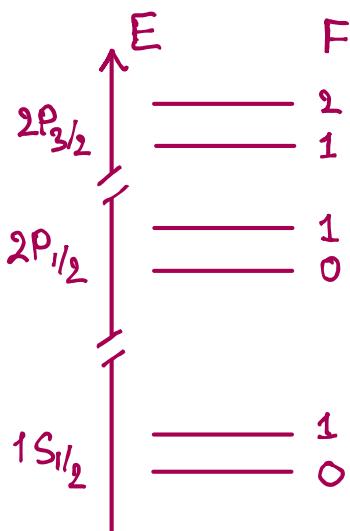
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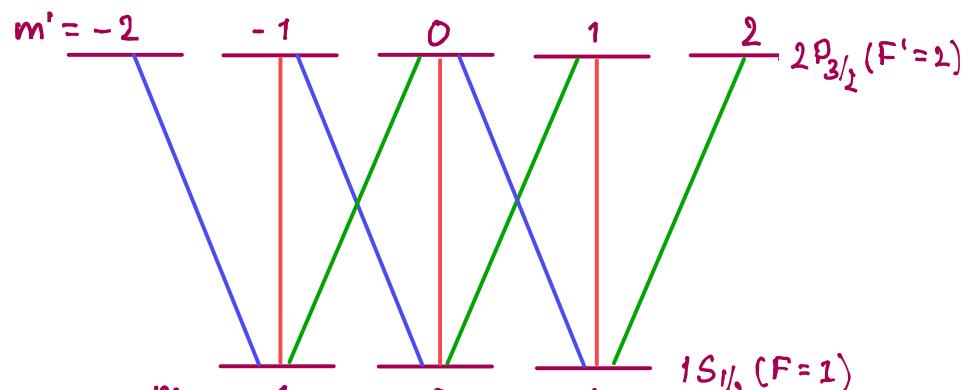
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Level diagram for transitions

$$1S_{1/2}(F=1) \rightarrow 2P_{3/2}(F=2)$$



Polarization:

$$q=0 \quad / \quad q=1 \quad \backslash \quad q=-1$$

Note: When the field polarization is pure linear or circular the levels are coupled in pairs, and the oscillator strengths depend on the Clebsch-Gordan coefficients

Demo: Clebsch-Gordan Coefficients and Oscillator Strengths from Mathematica

* Dense or hot gases: Collisions redistribute Atoms between m -levels on very short time scales and the gas looks like a gas of 2-level atoms w/an effective coupling strength. If the dipole is oriented at random with the field, Then $\langle \vec{r} \cdot \vec{\epsilon}_q \rangle_{\text{angles}} \sim \frac{1}{3} |\langle \vec{r} \vec{1} \rangle|$.

The same is true for unpolarized light

* Short interaction time: If the atoms are “unpolarized”(random m -level populations) and the interaction too brief to change this, the atoms behave as an ensemble with different oscillator strengths

* Optical pumping: In dilute gases without collisions, atoms can be “pumped” into a single, pure state, e. g., $1S_{1/2}(F=1, m_F=1)$. If driven with $\vec{\epsilon}_q = 1$ polarization this will realize a true 2-level system, as $2P_{3/2}(F'=2, m_F=2)$ can only decay back to $1S_{1/2}(F=1, m_F=1)$

* If more than one frequency or polarization is Present, one can often drive Raman transitions